## GYANMANJARI INSTITUTE OF TECHNOLOGY

## MECHANICAL ENGINEERING DEPARTMENT

## CLASS TEST No. 1

Subject: Elements of Mechanical Engg.
Date: 19.02.2016

## Instruction:

Marks: 30

1. All questions are compulsory
2. Make suitable assumption wherever necessary
3. Take $\mathbf{C p}=1.005 \mathrm{~kJ} / \mathrm{kgk} \mathbf{C v}=\mathbf{0 . 7 1 8} \mathrm{kJ} / \mathrm{kgk}$ for air

| Q.No | A | B | Mark |
| :---: | :---: | :---: | :---: |
| 1. | Explain with neat sketch the construction and working of Babcock and Wilcox boiler. | Explain with neat sketch construction and working of a Cochran boiler. | 07 |
| 2. | Derive PV/T=C with usual notation | Derive $\mathrm{Cp}-\mathrm{Cv}=\mathrm{R} \quad$ with usual notation | 07 |
| 3. | Explain Zeroth Law and First law of thermodynamics. | $\begin{array}{lll}\text { Define } & \text { Pressure and explain } \\ \text { Absolute } & \text { Pressure, Gauge Pressure }\end{array}$ and Atmospheric pressure | 04 |
| 4. | 5 kg of air is heated from an initial volume $0.5 \mathrm{~m}^{3}$ to final volume 1.3 $\mathrm{m}^{3}$ at constant pressure 4 bar Determine (i)heat supplied (ii) work done (iii)initial and final temperature of air | Air whose pressure, volume, and temperatures are 5.23 bar, $0.06 \mathrm{~m}^{3}$ and $196^{\circ} \mathrm{C}$ respectively has the state changed at constant pressure until its temperature becomes $27^{\circ} \mathrm{C}$. Calculate <br> (i) Heat Transferred. <br> (ii) Work Done during the process. | 07 |
| 5. | Write similarities and Dissimilar transfer. | ties between heat transfer and work | 05 |

